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ISOLATION AND CHARACTERIZATION OF CAFFEIC ACID FROM THYME SEEDS

RESEARCH ARTICLE

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ABSTRACT

Caffeic acid is a dihydroxycinnamic acid belonging to the phenolic acid family, which has a phenyl propanoidstructur. It occurs naturally in the wide range of food items such as vegetables, fruitsetc Thymespecies are widely used for medicinal purposes. Thyme seeds contains many flavonoids and several phenolic compounds. The present research article discuss about isolation of caffeic acid from thyme seeds and to elucidate its structure using UV, IR¹³, C-NMR, ¹ H-NMR AND MASS SPECTROSCOPY.

Key words: caffeicacid, thyme, flavonoids, isolation, elucidation, spectroscopy

INTRODUCTION

In the recent years ,there is an increased demands for the food which contain ingredients that may provide health benefits, besides nutrition., (Nanasombat etal). Thyme essential oil constitutes raw material in perfumery and cosmetics due to their aroma flavor (GrigoreA etal).

Thyme contains many flavonoids, phenolic anti-oxidants likezeaxanthin, lutein, pigenin, naringenin, luteolin and thymonin Fresh thyme is packed with minerals and vitamins that are essential for optimum health. thyme provides 0.35 mg of vitamin B-6 or pyridoxine furnishing about 27% of daily recommended intake. (Sharangi AB *et.al*)

Consumption of natural foods rich in flavonoids helps protect from lung and oral cavity cancer.(Aksel B)

Thyme have been used in traditional medicine for the treatment of several respiratory diseases like asthma and bronchitis (ocana A and reglero *et.al*). In addition to this, it has several properties, including, antiseptic, antispasmodic, anti tussive, anti oxidative, anti viral (Prashanth R etal) and anti microbial (Boruga *et.al*).

The present study aimed to isolate the polyphenol caffeic acid from thyme seeds and to elucidate its structure using various spectrometric methods.

Experimental procedure

Extraction

About 500 grams of the thyme seeds were soaked in a round bottom flask by using hydro alcoholic solution (70:30) for about 24 hours. The content of the round bottom flask was extracted by reflux for 2 hours. The extract was filtered off and filtrate was collected separately. The filtrate was treated with equal quantity of ether, ethyl acetate and n-butanol and each extracts were collected and concentrated separately.

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Purification by column chromatography

Mobile phase as chloroform: ethyl acetate in the ratio of 100, 90:10, 80:20, 70:30, 60:40, 50:50 were taken and Stationary phase as silicagel G. The concentrated organic layers were allowed to run through the column and different fractions were collected and are tested for the presence of flavonoids. The fraction 70:30 and 60:40 showed the positive test for the presence of flavonoids. These two layers were concentrated by using distillation (solvent recovery) and evaporated to get the crude caffeic acid.

Characterization of isolated compound Caffeic acid

Crystallized from MeOH as pale yellow needles, $C_9H_8O_4$, Melting point 210-212°C, gave brisk effervescence with saturated NaHCO₃ solution, light blue with Fe³⁺ and decolourised Br₂ water.

It was blue under UV and deep blue under UV/NH₃. Shows positive for shinoda test.

IRanalysis (γmax, cm⁻¹, KBr) (Figure-2.) 3427, 2558, 1654, 1605, 1534, 1449, 1283, 1217, 1176, 1110, 969, 901, 851, 809.

¹³C **NMR** (500MHz, DMSO-d₆, δ, ppm) (Figure-3) 168.51(s,>C=O); 148.64 (s, C-β); 146.05 (C-3); 145.16 (C-4); 126.19 (C-1); 121.79 (C-6); 116.25 (C-5); 115.59 (C-2); 115.0 (C-α).

¹**H NMR** (500MHz, DMSO-d₆, δ, ppm) (Figure- 4)7.38 (d, J=16.0 Hz, 1H, H-α), 6.99 (d, J=2.3 Hz, 1H, H-2), 6.92 (dd, J=8.4 & 2.3 Hz, 1H, H-6); 6.72 (d, J=8.4 Hz, 1H, H-5); 6.15 (d, J=16.05 Hz, 1H, H-β) **MASSSPECTRA** (Figure: 5)180 (M⁺, 100); 181 (M+H); 179 (M-H)

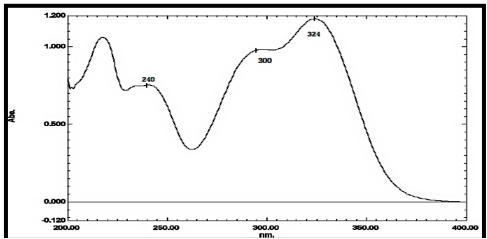


Figure-1: UV spectrum of the compound

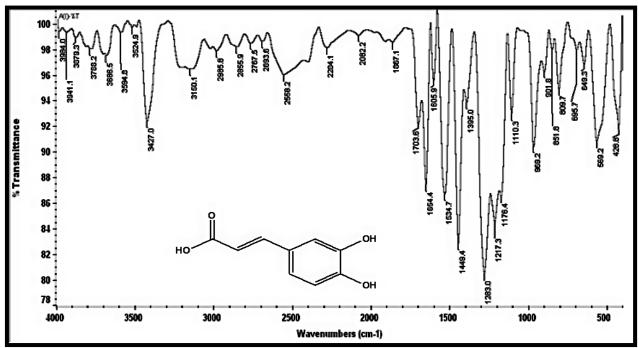
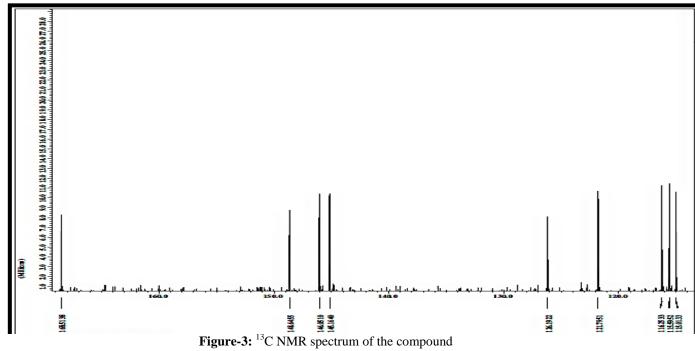


Figure-2: IRspectrum of the compound



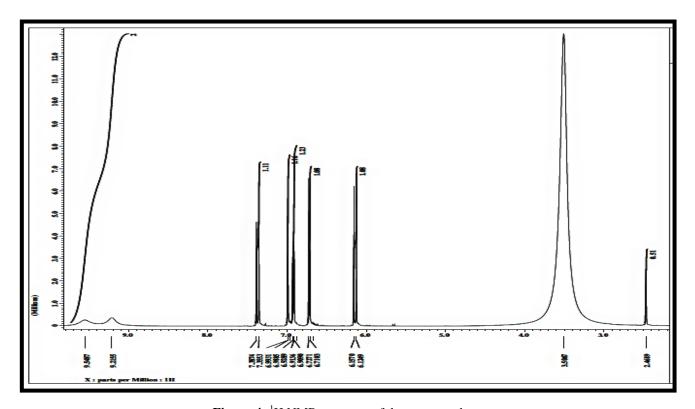


Figure-4: ¹H NMR spectrum of the compound

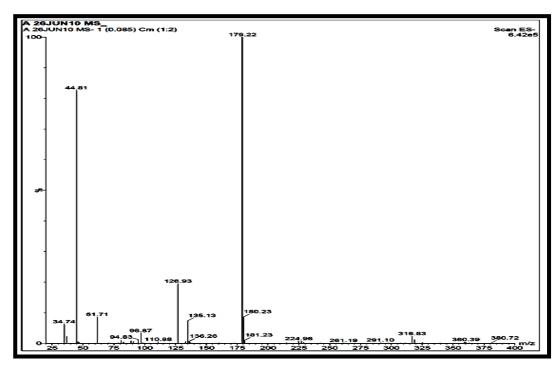


Figure -5: Mass spectrum of the compound

RESULTS AND DISCUSSION

The crude caffeic acid is isolated from the thyme seeds and its yield was about 0.779 grams. The structural identification of the isolated compound was carried out by MASS SPECTROMETER, ¹³C-NMR, ¹H-NMR, IR and UV SPECTROMETER. From the above given spectral data, it is identified as a caffeic acid.

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